

**WARM UP** All of the statements below are false as written.  
Write a term or phrase that makes the statement true when it is substituted for the underlined word(s).

1. Superscripts are used to show how many atoms of each element are found in a compound.
2. An ion is a(n) neutral particle that has either more or fewer electrons than protons.
3. A polyatomic compound contains two different elements.
4. The properties of a compound are the same as the properties of the elements that it contains.
5. When writing chemical formulas, add oxidation numbers so that the sum is equal to ten.