## Warm Up

## Name the compounds below.

Each of the cations below can have more than one oxidation # (think about Roman Numerals).

Determine the oxidation number for each cation and thus, determine which cation it is.

For instance...Tin IV has an oxidation # of +4

```
Pb(SO<sub>4</sub>)<sub>2</sub>
AsCl<sub>5</sub>
CuCl
Fe(OH)<sub>2</sub>
```