

Warm Up



Below are compounds with Transition Elements. Each of the cations below can have more than one oxidation #. Determine the oxidation number for each cation and thus, determine which cation it is. Write the name of the compound. Be sure to include the Roman Numeral in the name. For instance... Tin IV has an oxidation # of +4



$\text{Sn}(\text{SO}_4)_2$ is
Tin IV Sulfate