PS Chemistry Metric Practice	Name Period	
	tion for cost module and	
write the correct appreviation for each metric unit.		
1) Kilogram	4) Milliliter	7) Kilometer
2) Meter	5) Millimeter	8) Centimeter
3) Gram	6) Liter	9) Milligram
Try these conversions, using the ladder method.		
1) 2000 mg = g	6) 5 L = mL	11) 16 cm = mm
2) 104 km = m	7) 198 g = kg	12) 2500 m = km
3) 480 cm = m	8) 75 mL = L	13) 65 g = mg
4) 5.6 kg = g	9) 50 cm = m	14) 6.3 cm = mm
5) 8 mm = cm	10) 5.6 m = cm	15) 120 mg = g
Compare using <, >, or =.		
16) 63 cm 🔵 6 m	17) 5 g 🔵 508 mg	18) 1,500 mL①1.5 L
19) 536 cm 🔵 53.6 dm	20) 43 mg 🔵 5 g	21) 3.6 m 🔵 36 cm

Calculate Density.

Density is a derived unit because it is a combination of two other variables, mass and volume. When calculating density, mass or volume please show your work, round to the hundredths and label to receive full credit.

density =
$$\frac{\text{mass}}{\text{volume}}$$
 d = $\frac{\text{m}}{\text{v}}$

 A block of wood has a volume of 15 mL and a mass of 171 g. What is the density of the block of wood?

2) A rock is dropped into a graduated cylinder filled with water. The volume is measured to be 40 mL. The rock is then placed on a triple beam balance and the mass is measured to be 148 g. What is the density of the rock?

3) A block of wood has a density of 0.6 g/mL and a volume of 1.2 mL. What is the mass of the block of wood?

4) A 800 g rock has a density of 8 g/mL. What is the volume of the rock?



5) A student finds a rock on the way to school. In the laboratory he determines the volume of the rock. He fills a graduated cylinder with 100 mL of water. When he drops the rock into the graduated cylinder the water rises to 125.5 mL. What is the volume of the rock?

6) He then places the rock (question #5) on a triple beam balance and measures the mass of the rock. He records the mass as 427.3 g. What is the density of the rock?

7) The volume of a strangely shaped object is 22.7 mL and the mass is 39.9 g. What is the density of the object?